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completes each statement or best answers each question. _____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b.

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the sum of the partial pressures of the

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component gases. 2. Chapter Test B Gases
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(10) at constant volume and temperature
the total pressure exerted by a mixture of
gases is equal to the sum of partial pressures
of the component gases.

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(b) An English unit of pressure sometimes used in engineering is pounds per square inch ($\text{lb}/\text{in.}^2$), or psi: $1 \text{ atm} = 14.7 \text{ lb}/\text{in.}^2$. If a pressure is reported as 91.5 psi, express the measurement in atmospheres. Answer: (a) 99.3 kPa, (b) 6.22 atm

Practice Exercise 1) A gas at a pressure of 10.0 Pa exerts a force of

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_____ N on an area of A) 55

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Solids, Liquids, and Gases Chapter Test A

Multiple Choice Write the letter of the
correct answer on the line at the left. _____

1. The surface of water can act like a sort of
skin due to a property of liquids called a.
viscosity. b. surface tension. c.

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condensation. d. evaporation. _____ 2.

Which state of matter undergoes changes in volume most ...

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